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Multilayers of zinc-blende half-metals with semiconductors

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Abstract

We report on first-principles calculations for multilayers of zinc-blende half-metallic ferromagnets CrAs and CrSb with III–V and II–VI semiconductors, in the [001] orientation. We examine the ideal and tetragonalized structures, as well as the case of an intermixed interface. We find that, as a rule, half-metallicity can be conserved throughout the heterostructures, provided that the character of the local coordination and bonding is not disturbed. We describe a mechanism operative at the interfaces with semiconductors that can also give a non-integer spin moment per interface transition atom, and derive a simple rule for evaluating it.

1. Introduction

Half-metallicity is the property of some spin polarized materials of exhibiting a metallic density of states for one spin direction but a clear band gap around the Fermi level $E_{\rm F}$ for the other [1]. Well known half-metals include some Heusler alloys [1–4], the compound ${\rm CrO_2}$ [5], the manganite ${\rm La_{0.7}Sr_{0.3}MnO_3}$ [6], magnetite (Fe₃O₄) [7] and some diluted magnetic semiconductors (DMS) [8]. In recent years technological interest in possible applications in spin-dependent transport has fuelled the research in this field. For instance, ordered zinc-blende CrAs and CrSb have been fabricated by molecular beam epitaxy, and experiments and calculations suggest half-metallicity [9, 10]. The high Curie temperature $T_{\rm C}$ (over 400 K in experiment [11, 12] and in theory [13]) makes these compounds attractive also for applications. CrAs/GaAs [10] and CrSb/GaAs [14] multilayers have also been realized, showing that coherent heterostructures are possible.

Such zinc-blende compounds of transition elements with group V and VI atoms (called pnictides and chalcogenides) have also been studied by *ab initio* methods. Calculations

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of the total energy and structural optimization agree with the experimental result that the ground state structure of such compounds is not the zinc-blende one, but in most cases the NiAs structure [15]—rather more complicated and certainly not half-metallic [16–20]. The zinc-blende structure can only be achieved by epitaxial growth on semiconductors for a few monolayers. Heterostructures of half-metals with semiconductors are technologically interesting, since in principle they can be used to achieve spin injection from the ferromagnetic electrode to the semiconductor with 100% spin polarized current, a very useful quality for potential applications in spin transistors [21].

Until now, among the high $T_{\rm C}$ half-metals, heterostructures of Heusler alloys with semiconductors have been found to suffer from interface states within the half-metallic gap [22, 23], which can strongly reduce the spin polarization of the current. On the other hand, zinc-blende half-metals follow the structure and the bonding of the zinc-blende semiconductors in a more coherent way, and thus one can expect to avoid spurious interface states. Motivated by these ideas, we investigate in this paper the half-metal/semiconductor (HM/SC) contact for the zinc-blende materials and see under which conditions half-metallicity is preserved; moreover, we investigate the magnetic moments at the interface.

The paper is organized as follows. In section 2 we give a few details on our calculations. In section 3 we briefly review the bulk properties of these half-metallic compounds, and address the effect of lattice distortion by tetragonalization. In section 4 we present calculations on HM/SC multilayers in ideal geometry and epitaxy and see in which cases half-metallicity is preserved. In section 5 we examine interdiffusion effects at the interface and in section 6 we discuss the spin magnetic moments at the interface. Finally, we conclude with a summary in section 7.

2. Method of calculation

Our calculations are based on density functional theory within the local density approximation (LDA) for the exchange–correlation potential, with the Vosko, Wilk and Nusair parametrization [24]. We employ the full-potential screened Korringa–Kohn–Rostoker (KKR) Green function method [25], where the correct shape of the Wigner–Seitz cells is taken [26]. For the calculation of the charge and spin density we integrate along a contour on the complex energy plane, which extends from below the valence s states of the sp atom up to the Fermi level, using 42 energy points; the lower states are treated as atomic core states. For the Brillouin zone (BZ) integration we have used a **k**-space grid of $30 \times 30 \times 30$ in the full BZ for the bulk calculations and a 20×20 **k** $_{\parallel}$ -space grid in the two-dimensional full BZ for the interface calculations. We have used a cut-off of $\ell_{max}=3$ for the wavefunctions and Green functions. More details are given in [27]. The calculations are performed within the scalar relativistic approximation, which accounts for relativistic effects other than the spin–orbit coupling. The effect of the latter on the half-metallic property is quite small for CrAs and CrSb, as shown in [28] (the reduction of the spin polarization at the Fermi level is of the order of 0.5% for CrAs and 1.5% for CrSb).

Finally, throughout the paper we have used the experimental lattice parameters for the semiconductors (see [29–31] for a review).

3. Bulk properties

3.1. Ideal zinc-blende structure

The bulk properties of zinc-blende half-metals have been discussed in many papers in the past [17–20, 27, 32–42]. Here we will summarize what is already known, and use this as a

step towards the understanding of the bulk properties under tetragonal distortion and of the interface properties.

In the zinc-blende structure every atom has tetrahedral coordination, with its first neighbours being of the other atomic species. This symmetry splits the d states of the transition-metal element (TM) into two subspaces: the t_{2g} with threefold degeneracy (d_{xy} , d_{yz} and d_{xz} orbitals) and the e_g with twofold degeneracy (d_{z^2} and $d_{x^2-y^2}$). The t_{2g} states of the TM hybridize with the p states of the neighbouring sp atom (group V or VI), forming hybrids of bonding (lower in energy) and antibonding (higher) nature. This hybridization and bonding–antibonding splitting is crucial for the formation of the gap in these compounds, and it is a characteristic of the tetrahedral coordination—in the hexagonal coordination of the energetically more stable NiAs structure, for instance, it is not present. The effect is already known as 'p–d repulsion' in TM impurities in zinc-blende semiconductors [43].

The hybrids form wide bands, which retain the bonding–antibonding character and are therefore accordingly separated in energy. On the other hand, the e_g orbitals are practically non-bonding and form narrow bands, which remain energetically between the bonding and antibonding t_{2g} states, separated from both. Meanwhile, the s states of the TM are still higher in energy, pretty much like the s state of the Ga atom in GaAs stays above E_F . Only in cases of heavy sp anions, which have reduced ionicity, do these s states come closer to E_F but still stay above it for minority spin—see for instance a recent study on MnBi [44]. The s states of the sp anion, on the other hand, are anyway very low, lower than the bonding p–d hybrids.

This energetical separation already creates a band gap, and is assisted strongly by the exchange splitting due to spin magnetism. Thus, after the low lying bonding bands are filled, the e_g bands of only majority spin are progressively filled, while the corresponding minority bands are pushed higher above E_F by the exchange splitting. As we change the valence of the TM atom toward the right in the periodic table, more majority bands are filled; the magnetic moment increases; thus the exchange splitting also increases keeping the minority e_g states high in energy. In this way E_F remains between the well separated bonding bands and the e_g bands, within a band gap. As a typical example we present the band structure of CrSb in figure 1 (left) for energies around the gap.

The situation described above changes when the lattice is compressed. Then the extended majority spin states of p character feel the volume reduction and push $E_{\rm F}$ higher, so in the end $E_{\rm F}$ enters the minority spin conduction band. After this point, although the gap still exists, it is below $E_{\rm F}$ and half-metallicity is lost. Therefore it is necessary to identify which compounds are half-metallic in their equilibrium lattice parameter, and of course on which semiconductors they might grow. In [27] we have identified such systems, and we summarize the most important results here in table 1.

3.2. Tetragonal distortion of the zinc-blende structure

When a zinc-blende half-metallic compound is grown on a semiconductor substrate (say on (001)), its lattice will be tetragonally distorted so that it can assume the in-plane lattice constant of the SC while approximately keeping its own atomic volume. If the lattice match is good enough, the distortion will be minimal, and the electronic structure will change only slightly. Here we shall investigate this effect for the case of reasonable HM/SC lattice matching, in order to show that half-metallicity is not destroyed by moderate tetragonalization.

As a model system for the discussion we choose CrSb, which we assume to be deposited on ZnTe (001) and to have taken the corresponding in-plane lattice parameter of the semiconductor $a_{\text{ZnTe}} = 6.1 \text{ Å}$ and otherwise to have kept the atomic volume. This gives a ratio c/a = 0.914, where c is the lattice parameter in the growth direction. The corresponding band structure is shown in figure 1 (right). Comparing to the CrSb bands in the ideal zinc-blende geometry

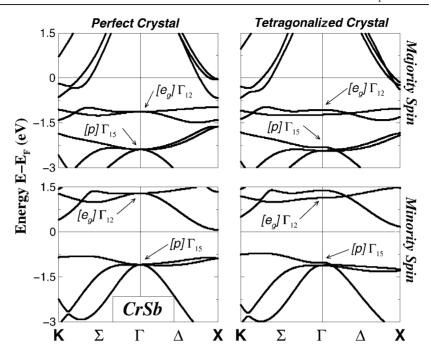


Figure 1. Left: the band structure of zinc-blende CrSb at the equilibrium lattice parameter of 5.92 Å. Right: the same system, but with a moderate tetragonal distortion along the z axis to obtain the ZnTe lattice constant in the (001) plane while keeping the equilibrium atomic volume. The triply degenerate t_{2g} states at Γ has split into two subspaces, one singly (p_z-d_{xy}) and one doubly degenerate (p_z-d_{yz}) and (p_z-d_{yz}) . Similarly, the doubly degenerate (p_z-d_{yz}) and (p_z-d_{yz}) have split into two subspaces. (Since the symmetry is no longer cubic, the group representations at Γ are no longer (p_z-d_{yz}) and (p_z-d_{yz}) have split into two subspaces. (Since the symmetry is no longer cubic, the group representations at Γ are no longer (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) are no longer cubic, the group representations at Γ are no longer (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) (p_z-d_{yz}) are (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) are (p_z-d_{yz}) and (p_z-d_{yz}) are (p_z-d_{yz}) and (p_z-d_{yz}) are (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) are (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (p_z-d_{yz}) and (

Table 1. Zinc-blende compounds that are half-metallic (HM) at their equilibrium lattice parameter a (calculated within the LDA) together with semiconductors (SC) with lattice parameter close to that value (experimental values given). A + means that the compound is HM at the lattice constant of the given SC, and a + means that the lattice mismatch is small, giving a candidate for epitaxial growth. For more details see [27].

НМ	SC a (Å)	GaAs 5.65	CdS 5.82	CdSe 6.05	InAs 6.06	GaSb, ZnTe 6.10
VAs	5.65	+	+	+	+	+
VSb	5.98	_	+	+	+	+
CrAs	5.52	+	+	+	+	+
CrSb	5.92	_	+	+	+	+
CrSe	5.61	_	+	+	+	+
CrTe	6.07	_	_	+	+	+

(figure 1 (left)), we see that the threefold degeneracy of the t_{2g} states at Γ has split into two subspaces, one singly (p_z-d_{xy}) and one doubly degenerate (p_x-d_{yz}) and p_y-d_{xz} . Also the e_g representation at Γ has split into its constituents, d_{z^2} and $d_{x^2-y^2}$. This behaviour is expected because the z axis is now distinguished. Evidently half-metallicity is preserved by this moderate distortion.

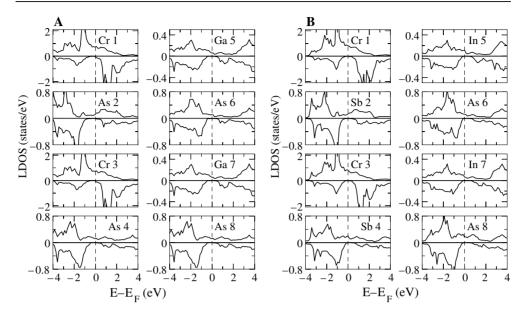


Figure 2. The atom-resolved DOS of CrAs/GaAs (A) and CrSb/InAs (B) (001) multilayers in supercell geometry. The inset numbers refer to the enumeration of successive layers. In CrAs/GaAs, the following atoms have equivalent environments and DOS: Cr1 and Cr3, As4 and As8, Ga5 and Ga7.

Such behaviour is typical for all HM examined here. The tetragonalization can have an effect on half-metallicity if $E_{\rm F}$ is at the edge of the gap for the ideal structure. Then mainly the extended p states, feeling the lateral change, will push $E_{\rm F}$ higher, similarly to the case of lattice compression; ultimately, if the distortion is too large, half-metallicity can be destroyed. In particular, if c/a < 1 the p_z states are squeezed and move higher in energy, while if c/a > 1 this happens to the p_x and p_y states.

4. HM/SC multilayers

In a recent examination of possible combinations of the transition elements V, Cr and Mn, with group V and VI elements [27], we found that some of them are half-metallic at their equilibrium lattice constants, which also fit reasonably well to those of some semiconductors (SC). Here we examine the cases of CrAs/SC and CrSb/SC (001) multilayers for which the lattice mismatch should be small, as can be seen from table 1. Note that within the LDA the lattice constant is usually underestimated by up to 2–4% [45]. Thus, if one allows for such a small increase of the calculated values of table 1 for the HM lattice constants, plus a moderate adjustment due to lattice mismatch, one arrives at the HM/SC combinations examined here.

We assume that the structures have the experimental SC lattice constant and that the ideal zinc-blende structure is kept throughout. The systems consist of four monolayers (ML) of HM followed by 4 ML of SC and periodically repeated, in accordance with the experimental result that only a few ML can exist within the periodically repeated supercell in the CrAs/GaAs case [10]. In the [001] direction of growth, this corresponds to interchanging monatomic layers: e.g., for CrAs/GaAs, we have a supercell of the form... Cr/As/Cr/As/Ga/As/Ga/As.... For test purposes we have performed calculations on 8 ML HM/8 ML SC, and seen that our conclusions remain unchanged.

Our results on the local DOS (LDOS) are presented in figure 2 for CrAs/GaAs (left) and for CrSb/InAs (right). In the multilayers we see that half-metallicity is conserved throughout.

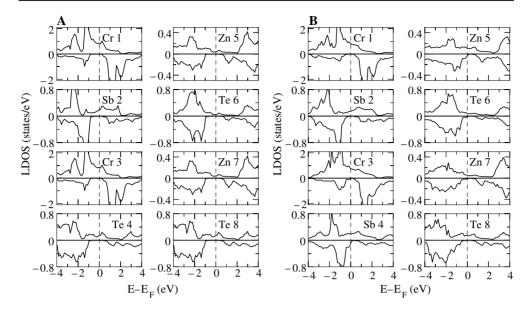


Figure 3. The atom-resolved DOS of the CrSb/ZnTe(001) multilayer in supercell geometry, in two different configurations. (A) a multilayer of the form Cr/Sb/Cr/Te/Zn/Te/Zn/Te. Half-metallicity is present throughout. (B) a multilayer of the form Cr/Sb/Cr/Sb/Zn/Te/Zn/Te, where the difference from case (A) is that here a Sb/Zn contact is present. This destroys half-metallicity, as an interface state is formed at the Sb/Zn interface for the minority spin states. This can be seen in particular at the Sb (4) and Zn (5) atom LDOS.

This means that no interface states within the gap are formed for the minority spin at the HM/SC interface. This can be understood since the growth is coherent, so the local environment of the interface Cr atoms is not changed in CrAs/GaAs (this is also the case in CrAs/MnAs multilayers; see, e.g., [35]). In CrSb/InAs each interface Cr atom has two Sb neighbours on the one side and two As neighbours on the other, but the minority gap remains, since the p-d hybridization and the bonding-antibonding splitting are still realized (the difference of having an As neighbour instead of an Sb one is that the p states of the former are somewhat lower in energy).

For the majority spin the local DOS at $E_{\rm F}$ decays within the SC layers. For a thick SC spacer it should vanish far from the interface, since the SC band gap is present and the DOS comes from exponentially decaying metal-induced gap states.

The main difference between the CrAs/GaAs multilayer and the CrSb/InAs one is in the position of $E_{\rm F}$ within the minority gap. In CrAs/GaAs, $E_{\rm F}$ just touches the conduction band, while for CrSb/InAs it is lower, closer to the middle of the gap. This is a volume effect, clear also in the bulk CrAs case. The reduced lattice constant of CrAs/GaAs compared to CrSb/InAs affects the position of the $E_{\rm F}$ within CrAs: the p electrons feel the reduced volume and shift towards higher energy, also pushing $E_{\rm F}$.

Next we present the multilayer CrSb/ZnTe. We can call this a 'mixed valence' multilayer, because it combines a 3d/group V HM with a II/VI semiconductor. Thus, the environment of the interface Cr atom is highly anisotropic: on the one side it has two Sb neighbours and on the other two Te neighbours, the electronegativity of which is higher than that of Sb. Nevertheless, half-metallicity survives, since the p-d repulsion creating the bonding-antibonding splitting is still there (although with different Cr–Sb and Cr–Te strengths), and the strong exchange splitting is of course also present. The half-metallic LDOS is shown in figure 3 (left).

However, there are two inequivalent ways to construct the multilayer, accounting for two different possible interfaces. The first is Cr/Sb/Cr/Te/Zn/Te/Zn/Te, which is the half-metallic one presented in figure 3 (left), and the second is Cr/Sb/Cr/Sb/Zn/Te/Zn/Te, the difference being that here a Sb/Zn contact is present, i.e., a III-VI hybrid. In the latter case half-metallicity is destroyed, due to interface states which occur at the Sb/Zn interface at E_F for the minority spin LDOS. The corresponding LDOS is shown in figure 3 (right), where the minority spin interface states are formed at E_F for the Sb (4) and Sb (5) atoms. Similarly, only the Cr/Te/Cr/As/In/As/In/As stacking would retain half-metallicity in the case of a CrTe/InAs system, since in the case of the Cr/Te/Cr/Te/In/As/In/As stacking an In/Te interface is present.

Closing this section, we note that in the case of interfaces of half-metallic Heusler alloys with SC, the half-metallic gap is usually destroyed [22, 23, 46], e.g. from all the (theoretically) studied cases in [22] only the NiMnSb(111)/CdS(111) interface retained the half-metallicity, and this only when the interface was between the Sb and S atoms. Here, on the contrary, the HM gap is, as a rule, present, since the nature of the bonding is continued coherently at the interface.

5. Intermixed interface

Up to now we have considered only the ideal interface, without interdiffusion. However, in an experiment one can imagine that some intermixing can occur, and the question of keeping half-metallicity is raised again. As before, this can be discussed in terms of the bonding character.

Consider, for instance, the CrAs/GaAs(001) interface, with an intermixed CrGa monolayer (i.e., a layer sequence of ... Cr/As/Cr_{0.5}Ga_{0.5}/As/Ga/...), under the assumption that the tetrahedral cation—anion coordination is not severely distorted (this is a necessary requirement in all considerations of half-metallicity in these materials). Then, the first neighbours of each interface Cr atom will be As atoms, and the same holds for the neighbours of the Ga atoms. A Cr–Ga bond will not be present, since the closest distance between Cr and Ga atoms is the one of second neighbours. Thus we expect the same bonding—antibonding scheme as before, so the HM gap should remain. Changes *will* occur, of course, for example due to the different electrostatic potential at the interface, since Cr and Ga gave different ionicities. Thus $E_{\rm F}$ can be shifted with respect to the ideal interface.

In order to investigate whether the HM property is really preserved, we performed a calculation for such an intermixed interface, by constructing an in-plane supercell, double the size of the usual cell and containing one Cr and one Ga atom at the interface. The DOS in the vicinity of $E_{\rm F}$ is shown in figure 4. Evidently, the system is half-metallic.

6. Magnetic moments

In the bulk HM, the total spin moment per unit cell is integer (in μ_B), as follows from just noticing that, due to the gap at E_F , the number of minority spin electrons is an integer. This line of thinking leads to a simple 'rule of 8' [27] connecting the total number of valence electrons Z_{tot} of the unit cell and the total spin moment M_{tot} per unit cell:

$$M_{\text{tot}} = (Z_{\text{tot}} - 8) \,\mu_{\text{B}},\tag{1}$$

since one has a total of eight electrons in the bonding p-d bands and in the (deeper lying) s band of the sp element, and the remaining the electrons occupy only majority states and build up the magnetic moment. This 'rule of 8' is generally valid for the whole family of zinc-blende half-metallic compounds, expressing a Slater-Pauling behaviour; similar rules

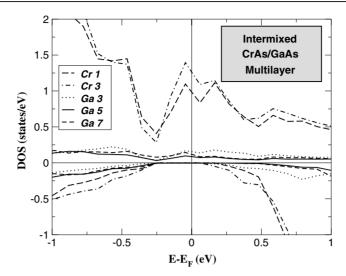


Figure 4. The atom-resolved DOS of the CrAs/GaAs(001) multilayer in the supercell geometry, in the case of an intermixed interface. Along the growth axis the successive layers are $Cr/As/Cr_{0.5}Ga_{0.5}/As/Ga/As/Ga/As$; thus in the third layer we consider both Ga and Cr sites (denoted as Cr3 and Ga3 in the figure labels). The system is half-metallic.

hold also for the Heusler alloys [2–4] or HM superlattices [35]. As regards the distribution of the spin moment among the atoms, the sp element has an induced moment opposite to that of the transition element (thus for the local moment of the transition atom we have $M_{\rm loc}^{\rm TM} > M_{\rm tot}$). This can be understood in terms of the local character of the occupied bonding p–d majority and minority bands (majority ones are more d-like around the transition atom; minority ones are more p-like around the sp atom). For details we refer the reader to [27]. Here we discuss the magnetic moments of the multilayers and interfaces.

Since at the interface the translational symmetry is broken, the magnetic moment per transition atom no longer has to be an integer, even if half-metallicity is preserved. This has been observed already in the case of (001) surfaces [27]. In the case of a transition-atomterminated interface with a SC, the total moment per interface transition atom depends on the valency of its neighbours on each side. To understand this, it is convenient to consider the occupied bonding p-d hybrids plus the low lying s states of the sp atom as a kind of 'reservoir' which has space for 6 + 2 = 8 electrons (both spin directions are occupied), and to count the electrons which fill it up; once the state reservoir is filled, the remaining electrons will occupy only majority spin states (e_g and p-d antibonding) and build up the moment. Every sp neighbour of the TM thus introduces eight low lying states, and at the same time Z_{neighb} valence electrons; this adds up to a total of $8 - Z_{\text{neighb}}$ unfilled reservoir states, or holes, for absorbing the TM electrons. On the other hand, each one of these sp neighbours has a total of four neighbours itself; thus it contributes only 1/4 of its states to the particular TM. Adding these up, we find a modified 'rule of 8', applicable for the total spin moment per interface or surface TM in zinc-blende half-metals provided that half-metallicity is preserved at the interface or the surface (equation (1) is a special case of this):

$$M_{\text{interf/surf}} = \left(Z_{\text{TM}} + \sum_{\text{neighb}=1}^{N} \frac{(Z_{\text{neighb}} - 8)}{4}\right) \mu_{\text{B}}.$$
 (2)

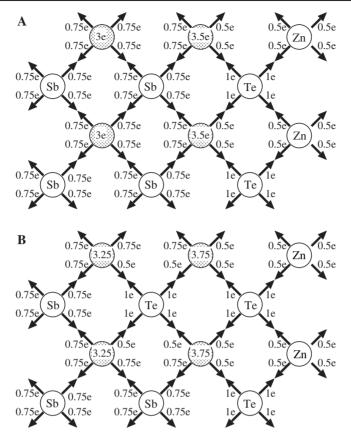


Figure 5. A schematic description for the non-integer moment at the interface ZnTe/CrSb. Shaded circles stand for Cr atoms. Arrows indicate the electrons donated by each atom to the bonding p–d bands. The remaining electrons at the Cr atoms occupy only majority spin states and build up the magnetic moment. In the bulk, one has 3 $\mu_{\rm B}$ per Cr atom. (A) an abrupt interface, where each interface Cr atom has two Sb and two Te neighbours. Since each Te atom donates an extra electron compared to the Sb atom, the interface Cr atom keeps an extra 1/2 electron and increases its moment by 0.5 $\mu_{\rm B}$. (B) an intermixed interface. The interface Cr atom has one Sb and three Te neighbours, while the sub-interface Cr atom has three Sb and one Te neighbours. The interface and sub-interface Cr spin moments are increased by 0.75 $\mu_{\rm B}$ and 0.25 $\mu_{\rm B}$, respectively, compared to the bulk CrSb value.

Here, N is the number of neighbours, which will be less than 4 in the case of a surface TM. As in the case of equation (1), this simple rule applies only when half-metallicity is present. The moment $M_{\text{interf/surf}}$ found by this rule is not expected to be completely localized at the TM, but rather distributed among the TM and the surrounding sites.

The situation is illustrated schematically in figure 5(A) for the case of ZnTe/CrSb (in the figure we do not consider the s states of the Sb and Te atoms, since they are energetically low lying and already filled by two electrons). In the bulk, each Sb neighbour of the Cr atom gives 3/4 of an electron to the p—d bond, so the four Sb neighbours give a total of three electrons per Cr atom. At the ZnTe/CrSb interface the Cr atom has two Sb and two Te neighbours. The Sb ones give again 3/4 of an electron each, while the Te give one electron each (since they have one more valence electron). Adding these up, we have a total of 3.5 e donated by the sp atoms to partly fill the bonding states of the interface Cr atom. Since there are six such

Table 2. Local spin moments (in units of μ_B) at the Cr and sp atoms in the CrAs/GaAs, CrSb/InAs and CrSb/ZnTe (half-metallic) (001) multilayers.

Layer	CrAs/GaAs		CrSb/InAs		CrSb/ZnTe	
1	Cr	2.930	Cr	3.195	Cr	3.389
2	As	-0.200	Sb	-0.253	Sb	-0.230
3	Cr	2.930	Cr	3.095	Cr	3.387
4	As	-0.107	Sb	-0.149	Te	-0.061
5	Ga	0.021	In	-0.005	Zn	0.022
6	As	-0.011	As	-0.056	Te	-0.007
7	Ga	0.021	In	-0.005	Zn	0.022
8	As	-0.107	As	-0.194	Te	-0.061

states, the remaining 2.5 e are donated by the Cr atom, while the magnetic moment is built up by the $Z_{\rm Cr}-2.5=3.5$ remaining electrons of Cr, where $Z_{\rm Cr}=6$ is the valence of Cr. As a result, a half-integer moment of $M=3.5~\mu_{\rm B}$ per interface Cr atom occurs. In fact, this cannot be strictly localized at the interface, but is distributed among the interface atoms and their neighbours.

Note that the above sum rules refer only to the total moments, and not to the local ones. For instance, table 2 gives the local spin moments for the half-metallic multilayers. In comparing CrAs/GaAS (total interface moment: 3 μ_B) with CrSb/ZnTe (total interface moment: 3.5 μ_B) results we find that the local interface Cr moment increases by about 0.45 μ_B and the other 0.05 μ_B is achieved by a reduction of the absolute value of the negative anion moments; interface Te atoms have a spin moment of around -0.06 μ_B compared to the -0.10 μ_B of the As interface atoms. In the case of the CrSb/InAs system the total interface moment should be 3 μ_B as for CrAs/GaAs and the larger Cr moment at the interface for the former system is compensated by the larger absolute values of the spin moments of the sp anions. The difference between the two systems arises mainly due to the larger electronegativity of As compared to Sb. The As p states are originally lower than the Sb ones, so the p-d hybridization is weaker, and the bonding states of majority spin are more localized around the As atom; thus the absolute values of both Cr and As moments are smaller.

But it is also possible to obtain a value of 3.25 $\mu_{\rm B}$ or 3.75 $\mu_{\rm B}$ per Cr atom. Such an example is shown schematically in figure 5(B), where we have substituted for the pure Sb layer at the interface a Sb_{0.5}Te_{0.5} intermixed layer. Each Cr atom at the interface has one Sb and three Te atoms as first neighbours, and gives away 2.25 electrons to the bonding states. Thus $Z_{\rm Cr}-2.25=3.75$ electrons are left to build up the spin moment. Similarly the Cr atoms in the sub-interface layer now have three Sb and one Te atoms as first neighbours and a total spin moment of 3.25 $\mu_{\rm B}$.

The above sum rule can be considered as a generalization of the sum rule derived in [27] for the surfaces. In the case of the Cr-terminated CrAs(001) or CrSe(001) surfaces, the Cr atom at the interface loses two of its four neighbours and its spin moment is increased by 1.5 μ_B in the case of the CrAs(001) surface and by 1.0 μ_B for the CrSe(001) surface.

7. Summary and conclusions

We have performed first-principles calculations for (001) multilayers of half-metallic zincblende compounds (CrAs and CrSb) with III–V and II–VI semiconductors, focusing on the question of whether half-metallicity is conserved at the interface. We have found that this can be the case under not too restrictive assumptions. Basically, the important requirement is that the coordination of the transition metal does not change at the interface: it should have four sp neighbours of anionic (electronegative) character, as in the bulk. Then the bonding–antibonding splitting of the p–d hybrids is retained and half-metallicity is conserved. It is not important that all four neighbours be of the same type. Thus, even in the InAs/CrSb and ZnTe/CrSb interfaces one observes half-metallicity. For the same reason, the intermixed Cr–Ga interface is half-metallic, too. We have also found that a moderate tetragonalization does not affect the half-metallic character of the materials; thus growth of these half-metals on semiconductors with slightly different lattice constant brings no problems.

We have also examined the magnetic moments at the interface. We have found that, in the case of an interface between a TM/group V compound and a II–VI semiconductor, the total spin magnetic moment at the interface has a non-integer value. On the basis of simple arguments, applicable to any zinc-blende ferromagnetic surface or interface provided that half-metallicity is preserved, we have derived a sum rule (equation (2)) for calculating this spin moment.

We conclude that zinc-blende half-metallic compounds combine various qualities desirable for spintronics applications: half-metallic ferromagnetism, high Curie temperatures, coherent growth and half-metallic interfaces with semiconductors. We hope that our work will motivate further experimental research in the field.

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References

- [1] de Groot R A, Mueller F M, van Engen P G and Buschow K H J 1983 Phys. Rev. Lett. 50 2024
- [2] Galanakis I, Dederichs P H and Papanikolaou N 2002 Phys. Rev. B 66 134428
- [3] Galanakis I, Dederichs P H and Papanikolaou N 2002 Phys. Rev. B 66 174429
- [4] Galanakis I 2004 J. Phys.: Condens. Matter 16 3089
- [5] Schwarz K 1986 J. Phys. F: Met. Phys. 16 L211 Korotin M A, Anisimov V I, Khomskii D I and Sawatzky G A 1998 Phys. Rev. Lett. 80 4305
- [6] Soulen R J Jr, Byers J M, Osofsky M S, Nadgorny B, Ambrose T, Cheng S F, Broussard P R, Tanaka C T, Nowak J, Moodera J S, Barry Q and Coey J M D 1998 Science 282 85
- [7] Yanase A and Siratori K 1984 J. Phys. Soc. Japan 53 312
- [8] Akai H 1998 Phys. Rev. Lett. 81 3002
- [9] Akinaga H, Manago T and Shirai M 2000 Japan. J. Appl. Phys. 39 L1118
- [10] Mizuguchi M, Akinaga H, Manago T, Ono K, Oshima M and Shirai M 2002 J. Magn. Magn. Mater. 239 269
- [11] Zhao J H, Matsukura F, Takamura K, Abe E, Chiba D and Ohno H 2001 Appl. Phys. Lett. 79 2776
- [12] Mizuguchi M, Akinaga H, Manago T, Ono K, Oshima M, Shirai M, Yuri M, Lin H J, Hsieh H H and Chen C T 2002 J. Appl. Phys. 91 7917
- [13] Kübler J 2003 Phys. Rev. B **67** 220403
- [14] Zhao J H, Matsukura F, Takamura K, Chiba D, Ohno Y, Ohtani K and Ohno H 2003 Mater. Sci. Semicond. Process. 6 507
- [15] Adachi K and Ogawa S (ed) 1988 Landolt-Börnstein New Series Group III vol 27, Part a (Heidelberg: Springer)
- [16] Wei S H and Zunger A 1987 Phys. Rev. B 35 2340
- [17] Sanvito S and Hill N A 2000 Phys. Rev. B 62 15553
- [18] Continenza A, Picozzi S, Geng W T and Freeman A J 2001 Phys. Rev. B 64 085204
- [19] Zhao Y J, Geng W T, Freeman A J and Delley B 2002 Phys. Rev. B 65 113202
- [20] Xie W-H, Xu Y-Q, Liu B-G and Pettifor D G 2003 Phys. Rev. Lett. $\bf 91$ 037204
- [21] Datta S and Das B 1990 Appl. Phys. Lett. 56 665
- [22] Wijs G A and de Groot R A 2001 Phys. Rev. B 64 R020402

[23] Picozzi S, Continenza A and Freeman A J 2003 J. Phys. Chem. Solids 64 1697 Picozzi S, Continenza A and Freeman A J 2003 J. Appl. Phys. 94 4723 Debernardi A, Peressi M and Baldereshi A 2003 Mater. Sci. Eng. C 23 743

- [24] Vosko S H, Wilk L and Nusair N 1980 Can. J. Phys. 58 1200
- [25] Papanikolaou N, Zeller R and Dederichs P H 2002 J. Phys.: Condens. Matter 14 2799
- [26] Stefanou N, Akai H and Zeller R 1990 Comput. Phys. Commun. 60 231 Stefanou N and Zeller R 1991 J. Phys.: Condens. Matter 3 7599
- [27] Galanakis I and Mavropoulos Ph 2003 Phys. Rev. B 67 104417
- [28] Mavropoulos Ph, Sato K, Zeller R, Dederichs PH, Popescu V and Ebert H 2004 Phys. Rev. B 69 054424
- [29] Madelung O (ed) 1982 Landolt-Börnstein New Series Group III vol 17, Part a (Berlin: Springer)
- [30] Madelung O (ed) 1986 Landolt-Börnstein New Series Group III vol 22, Part a (Berlin: Springer)
- [31] Rössler U (ed) 2001 Landolt-Börnstein New Series Group III vol 41, Part A1a (Berlin: Springer)
- [32] Shirai M 2001 Physica E 10 143
- [33] Galanakis I 2002 Phys. Rev. B 66 012406
- [34] Zhang M, Hu H, Liu Z, Liu G, Cui Y and Wu G 2004 J. Magn. Magn. Mater. 270 32
- [35] Fong C Y, Qian M C, Pask J E, Yang L H and Dag S 2004 Appl. Phys. Lett. 84 239
- [36] Xie W-H, Liu B-G and Pettifor D G 2003 Phys. Rev. B 68 134407
- [37] Liu B-G 2003 Phys. Rev. B 67 172411
- [38] Pask J E, Yang L H, Fong C Y, Pickett W E and Dag S 2003 Phys. Rev. B 67 224420
- [39] Sanyal B, Bergqvist L and Eriksson O 2003 Phys. Rev. B 68 054417
- [40] Xie W-H and Liu B-G 2003 J. Phys.: Condens. Matter 15 5085
- [41] Zhang M, Hu H, Liu G, Cui Y, Liu Z, Wang J, Wu G, Zhang X, Yan L, Liu H, Meng F, Qu J and Li Y 2003 J. Phys.: Condens. Matter 15 5017
- [42] Picozzi S, Shishidou T, Freeman A J and Delley B 2003 Phys. Rev. B 67 165203
- Zunger A and Lindefelt U 1983 Phys. Rev. B 27 1191
 Singh V A and Zunger A 1985 Phys. Rev. B 31 3729
 Wei S H and Zunger A 1988 Phys. Rev. B 37 8958
- [44] Xu Y G, Liu B G and Pettifor D G 2002 Phys. Rev. B 66 184435
- [45] Asato M, Settels A, Hoshino T, Asada T, Blügel S, Zeller R and Dederichs P H 1999 Phys. Rev. B 60 5202
- [46] Lezaic M, private communication